

Radical Alkynyltrifluoromethylation of Alkenes Initiated by an Electron Donor-Acceptor Complex

Heng Jiang, Yanyan He, Yuanzheng Cheng, and Shouyun Yu*0

State Key Laboratory of Analytical Chemistry for Life Science, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210023, China

Supporting Information

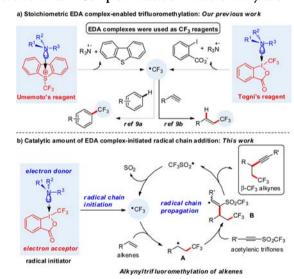
ABSTRACT: Radical alkynyltrifluoromethylation of alkenes with actylenic triflones has been achieved. This radical chain reaction is initiated by a catalytic amount of an electron-donor—acceptor complex composed of Togni's reagent and N-methylmorpholine. This transformation proceeds under exceptionally mild and operationally simple conditions. A variety of alkenes are compatible in this protocol including aliphatic alkenes, vinyl ethers, enecarbamates, styrenes, and even acrylates, providing diverse β -trifluoromethyl alkynes in good to excellent yields.



Radical addition reactions serve as an efficient and practical strategy to achieve difunctionalization of alkenes and alkynes through a radical-chain pathway. These reactions have been extensively studied through the development of diverse initiation conditions (ultraviolet irradiation, azo compounds, organotin reagents, or trialkylboranes). Recent progress focuses on the exploitation of highly efficient and environmentally benign visible-light photosensitizers to initiate the radical-chain addition reactions.² However, the scope of these visible-lightpromoted transformations is mainly restricted to the addition of an alkyl halide³ or trifluoromethylsulfonyl chloride (CF₃SO₂Cl)⁴ to an alkene to simultaneously generate a new C-C and C-X (X = halide or sulfur) bond. Therefore, the discovery of efficient and easily accessible initiators to achieve diverse radical-chain addition reactions,⁵ especially the difunctionalization of alkenes by simultaneously assembling two new C-C bonds across the C=C bond under mild conditions, is still challenging, but highly valuable in the radical chemistry community.⁶

Electron-donor-acceptor (EDA) complexes are a type of molecular aggregate generated from noncovalent interaction between an electron donor and an electron acceptor, which were disclosed by Mulliken decades ago. The EDA complex involved organic transformations related to single-electron-transferred radical pathways have gathered increasing attention with promising advances in recent years.8 Recently, our group defined a type of EDA complex composed of a tertiary amine with an electrophilic trifluoromethylating reagent (Togni's or Umemoto's reagent), which can be thermally activated to an excited state and subsequently dissociated into trifluoromethyl radical (°CF₃) and nitrogen-based radical cations. These EDA complexes had also been successfully applied to the trifluoromethylation of arenes and hydrotrifluoromethylation of alkenes and alkynes (Scheme 1a). Motivated by these achievements, we envisaged that the use of a catalytic amount of an EDA complex consisting of an electrophilic trifluoromethylating reagent and a tertiary amine would be capable of initiating a radical-chain addition of

Scheme 1. EDA Complex-Enabled Trifluoromethylation



acetylenic triflones to alkenes, leading to $\beta\text{-CF}_3$ alkynes in a single operation. 10

The rationale of this proposal originates from the generation of ${}^{\bullet}\text{CF}_3$ through single-electron transfer of an EDA complex composed of a tertiary amine and Togni's reagent (Scheme 1b). Radical attack of ${}^{\bullet}\text{CF}_3$ onto an olefin produces the radical intermediate **A**, which can be trapped by an acetylenic triflone to give radical intermediate **B**. β -Elimination of **B** affords triflyl radical (CF₃SO₂ ${}^{\bullet}$) and the desired alkynyltrifluoromethylation adduct of the alkene. Afterward, radical chain propagation is launched by reproducing ${}^{\bullet}\text{CF}_3$ through the release of SO₂ from CF₃SO₂ ${}^{\bullet}$. Although Fuchs and co-workers reported AIBN-

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Scheme 2. Scope of the Alkynyltrifluoromethylation of Alkenes with Silyl Alkyne-Derived Triflone $2a^{a,b}$

"Reaction conditions: 1 (0.2 mmol), 2a (0.24 mmol), Togni's reagent I (7.5 mol %), and NMM (10 mol %) in dry EtOAc (0.5 mL) at 30 °C for 24 h. "Isolated yield." dr = 1:1. "Adr > 20:1. "Reaction time: 48 h. "Idr = 12:1. "Reactions were performed in 0.5 mL of 1,4-dioxane for 72 h. "2 mmol scale."

initiated alkynyltrifluoromethylation of alkenes with actylenic triflones two decades ago, ^{10a} the strategy presented here possesses several interesting features: (I) a catalytic amount of EDA complex is used as an initiator, which avoids the use of a hazardous initiator; (II) this process simultaneously introduces two new C–C bonds; (III) acetylenic triflones serve as both trifluoromethyl and alkynyl sources, which makes this process atom economical; and (IV) trifluoromethyl radical (°CF₃) generated from the initiator is involved in the radical-chain propagation directly, which obviates the formation of byproducts derived from initiators.

This EDA complex initiated alkynyltrifluoromethylation of alkenes was initially examined through the reaction of phthalylprotected 1-penten-5-amine (1a) with silyl alkyne-derived triflone 2a (Scheme 2). A combination of 7.5 mol % of Togni's reagent I with 10 mol % of NMM in EtOAc was capable of initiating this transformation smoothly at 30 °C to give the desired product 3a in 92% yield. Togni's reagent II was ineffective in this transformation. Although the EDA complex of Umemoto's reagent III with NMM can be successfully used in C-H trifluoromethylation of arenes and hydrotrifluoromethylation of alkenes/alkynes, 9b it failed to initiate this reaction. Other electron donors including secondary and tertiary amines dramatically decreased the yield. Both of Togni's reagent I and NMM are indispensable; otherwise, the transformation could not be initiated with full recovery of starting materials (for full condition screening, see the Supporting Information).

Afterward, a variety of aliphatic alkenes were examined in this reaction (Scheme 2). Different kinds of functional groups were well tolerated under the optimal reaction conditions, including aromatic rings, heterocycles, silyloxy esters, free alcohols, carbamates, alkyl bromides, ketones, amides, aldehydes, esters,

and nitriles (3a-n). Cyclic alkenes (3o-q) and a sterically hindered alkene (3r) were also compatible in this reaction (Scheme 2a). It is noteworthy that vinyl ethers and enecarbamates proceeded smoothly in this reaction, providing the desired alkynyltrifluoromethylated adducts 3s-ab in good to excellent yields (Scheme 2b). Electron-neutral and electron-rich styrenes were also effective reaction partners with 2a, affording 3ac-ae in moderate yields (Scheme 2c).

We next sought to examine the reactivity of alkenes with phenylacetylene-derived acetylenic triflone 2b to prepare more structurally diverse β -CF₃ alkynes through this EDA complexinitiated radical reaction (Scheme 3). The combination of 5 mol % of Togni's reagent I and 10 mol % of NMM was capable of initiating this reaction efficiently to give the desired product 4a in 89% yield. A variety of functional groups in alkenes were well tolerated, including silyloxy esters, carbamates, ketones, esters, alkyl bromides, and amides, providing the desired alkynyltrifluoromethylation products in good to excellent yields (4b-g). Cyclohexene (4h), cycloheptene (4i) and the sterically hindered alkene (4j) also worked quite well in this transformation (Scheme 3a). Furthermore, vinyl ethers and enecarbamates were also suitable coupling partners with 2b, affording products 4k-pin satisfactory yields. Remarkably, electron-deficient alkenes such as acrylates and vinyl phosphonates were also effective to afford the desired products 4q-s in moderate to good yields (Scheme 3b). We also evaluated the reactivity of substituted styrenes with 2b (Scheme 3c). A variety of electron-rich or electron-deficient styrenes were well tolerated in this reaction, affording alkynyltrifluoromethylation adducts 4t-ah in moderate to satisfactory yields. Internal alkenes such as β -methylstyrene, α methylstyrene, indene, and 1,2-dihydronaphthalene were also compatible to produce 4ai-al in moderate to good yields. Some

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Scheme 3. Scope of the Alkynyltrifluoromethylation of Alkenes with Phenylacetylene-Derived Triflone $2b^{a,b}$

"Reaction conditions: **1** (0.24 mmol), **2b** (0.2 mmol), Togni's reagent **I** (5 mol %), and NMM (10 mol %) in dry EtOAc (0.5 mL) at 30 °C for 24 h.

^b Isolated yield.

^cdr = 1:1.

^ddr = 6:1.

^edr = 9:1.

^fdr > 20:1.

^gdr = 10:1.

^h10 mol % of Togni's reagent **I** was used, 48 h.

ⁱ10 mol % of Togni's reagent **I** was used in dry 1,4-dioxane (0.5 mL), 48 h.

^jdr = 4:1.

biologically important molecule-derived alkenes, such as estrone, δ -tocotrienol, (R)-limonene, and nootkatone, could also go through this ATRA reaction to give the corresponding alkynyltrifluoromethylation products 4am-aq in 71-85% yields (Scheme 3d).

The scope of arylacetylene-derived acetylenic triflones was also investigated (Scheme 4). Functional groups with various electronic properties, including OMe, t-Bu, Ph, Cl, Br, CF₃, and F, were well tolerated to furnish the desired products 5a—i in

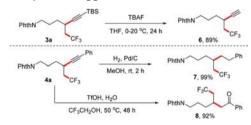
Scheme 4. Scope of Arylacetylene-Derived Triflones^{a,b}

^aReaction conditions: **1a** (0.24 mmol), **2** (0.2 mmol), Togni's reagent **I** (5 mol %), and NMM (10 mol %) in dry EtOAc (0.5 mL) at 30 $^{\circ}$ C for 24 h. ^bIsolated yield.

satisfactory yields (70–90%). Thiophene- and naphthalene-derived acetylenic triflones were also competent reaction partners with **2b**, affording **5j** and **5k** in 58% and 63% yields, respectively.

The resultant alkynyltrifluoromethylation products were easily modified (Scheme 5). For example, the silyl group of silylalkyne

Scheme 5. Synthetic Applications

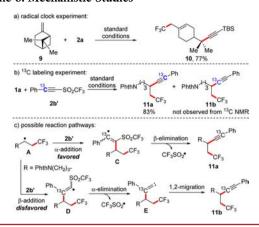


3a could be removed by treatment of **3a** with TBAF to give terminal alkyne **6**, which facilitates further transformations. The triple bond of alkyne **4a** could be easily reduced to compound 7 by Pd/C-catalyzed hydrogenation or converted to ketone **8** by acid-catalyzed hydrolysis. All of these transformations proceeded smoothly to give CF₃-containing synthetic building blocks in excellent yields, which could be potentially used in pharmaceutical chemistry for modern drug discovery.

To support the mechanism of the radical-chain-addition reaction we proposed in Scheme 1b, a series of control experiments were conducted (Scheme 6). The model reaction

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Scheme 6. Mechanistic Studies



could be totally terminated by introducing TEMPO into the reaction mixture. When (-)- β -pinene was employed under standard conditions, the ring-opening product 10 was obtained in 77% yield (Scheme 6a). These results suggest that the radical intermediates are involved in this transformation. Afterward, the ¹³C-labeled acetylenic triflone 2b' was used in this reaction (Scheme 6b). As a result, the ¹³C-labeled product 11a was isolated exclusively in 83% yield without detection of any 13Cmigration product 11b. This result supports the notion that α addition of radical intermediate A to acetylenic triflone 2b' occurs predominantly to afford vinyl radical C. β -Elimination of C produces 11a and CF₃SO₂. Therefore, the alternative pathway of β -addition and subsequent α -elimination to form the vinylidene carbene E followed by 1,2-migration to give 11b can be excluded in this radical-chain-addition reaction (Scheme 6c).¹³

In summary, we have described an electron donor—acceptor complex initiated radical-chain addition of acetylenic triflones to alkenes. The alkynyltrifluoromethylation of alkenes can be achieved under exceptionally mild and operationally simple conditions. The catalytic amount of EDA complex of Togni's reagent and NMM is capable of initiating this transformation efficiently. The reactivity of silylalkyne- and phenylacetylenederived acetylenic triflones with a variety of alkenes, including aliphatic alkenes, vinyl ethers, enecarbamates, styrenes, and even acrylates, was extensively discussed. A variety of structurally diverse β -CF₃ alkynes can be easily prepared in a single operation.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.7b00337.

Full experimental details, characterization of new compounds, and ^1H and ^{13}C NMR (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: yushouyun@nju.edu.cn.

ORCID [®]

Shouyun Yu: 0000-0003-4292-4714

Notes

The authors declare no competing financial interest.

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